

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.1.7.A 7	Unifying Themes	Explain the parts of a simple system and their relationship to each other.	<p>96 how systems may reach equilibrium</p> <p>97 equilibrium</p> <p>98 equilibrium</p> <p>99 equilibrium</p> <p>129 parts of a car and ramp system</p> <p>129 defining a system</p> <p>134 identify and describe parts of a system in nature</p> <p>134 understand concept of system</p> <p>219 restoring forces and equilibrium</p> <p>219 equilibrium and harmonic motion</p> <p>367 form and function</p> <p>390 recognize and describe how systems may reach equilibrium</p>	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.1.7.B 7	Unifying Themes	Describe the use of models as an application of scientific or technological concepts.	<p>featured in Ancillary Skillsheet "Dimensional Analysis", "Ratios and Proportions", "Gear Ratios"</p> <p>7 testing models</p> <p>36 using algebraic models</p> <p>48 design and test a model to solve a problem</p> <p>60 graphs</p> <p>61 making and evaluating graphs</p> <p>62 constructing a graph</p> <p>63 know that scientific knowledge can be in the form of models</p> <p>63 constructing graph from data</p> <p>83 using algebraic formulas</p> <p>86 using algebraic models</p> <p>98 using algebraic model</p> <p>115 using algebraic models</p> <p>124 design and test a model that solves a problem</p> <p>138 the power equation</p>	<p>9 calculate speed</p> <p>14 construct a graphical model</p> <p>15 construct a graphical model</p> <p>17 make graph from data</p> <p>22 make model from data</p> <p>27 calculate speed</p> <p>27 making a graph from data</p> <p>29 find the mechanical advantage</p> <p>56 find the mass/volume ratio</p> <p>83 scientific knowledge is sometimes in the form of models</p> <p>110 lab notebook</p> <p>111 making graphs</p>

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CPO Science *Physical Science (Middle School)***

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			150 design a model to solve a problem 200 equation for Ohm's law 222 harmonic motion graphs 227 calculating wave speeds 294 making a graph 325 design and test model 340 make and evaluate graphs	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.1.7.C 7	Unifying Themes	Identify patterns as repeated processes or recurring elements in science and technology.	156 orbits of planets and moons and other bodies 157 describe orbits of planets 158 orbit of moon 164 orbits of other bodies in the solar system 164 other bodies in solar system	32 orbit of the moon 65 identify groups or families on periodic table 67 identify metals and nonmetals and metalloids 67 recognizing groups or families
			165 compare Earth to other planets with respect to supporting life 166 compare planets to Earth 166 compare orbits of planets 166 classify and compare planets 167 classify and compare planets	
			168 compare other planets with Earth with respect to supporting life 168 classify planets 169 compare other planets to Earth with respect to supporting life 169 orbits of moons 169 classify planets	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
170			orbits of other moons	
170			classify and compare planets	
171			compare orbits of planets in solar system	
171			compare and classify planets	
172			classify planets	
173			orbits on bodies in solar system	
173			compare Earth to other planets with respect to supporting life	
173			classify planets	
286			Earth compared with other planets with respect to supporting life	
316			classify materials based on physical properties	
317			classify materials based on properties	
344			chemical and physical properties	
345			recognizing groups or families on the periodic table	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			<p>346 identify metals and nonmetals on the periodic table</p> <p>348 recognizing groups and families of periodic table</p> <p>348 recognizing metals and nonmetals and metalloids</p> <p>350 physical and chemical properties</p> <p>350 recognizing groups and families and periodic table</p> <p>353 groups on periodic table</p> <p>357 classify by chemical and physical properties</p> <p>357 melting and boiling points</p> <p>358 physical properties of gold</p>	
3.1.7.D 7	Unifying Themes	Explain scale as a way of relating concepts and ideas to one another by some measure.	<p>featured in Ancillary Skillsheet "Significant Differences in Measurement"</p> <p>12 rulers</p> <p>30 understand sensitivity of measuring tools</p> <p>76 rulers</p>	<p>16 understand the sensitivity of a measuring tool</p> <p>73 measure densities</p> <p>95 metric and English rulers</p> <p>96 metric rulers</p> <p>97 metric rulers</p> <p>98 metric rulers</p> <p>101 rulers</p>

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.1.7.E 7	Unifying Themes	Identify change as a variable in describing natural and physical systems.	<p>featured in Ancillary Skillsheet "Dimensional Analysis", "Ratios and Proportions", "Gear Ratios"</p> <p>4 speed</p> <p>6 speed</p> <p>34 speed</p> <p>35 understand and calculate speed</p> <p>36 using algebraic models</p> <p>40 understand concept of speed</p> <p>54 concept of speed</p> <p>55 speed</p> <p>56 speed</p> <p>60 calculating speed</p> <p>64 calculating speed</p> <p>75 speed</p> <p>76 speed</p> <p>83 using algebraic formulas</p> <p>86 using algebraic models</p> <p>98 using algebraic model</p> <p>115 using algebraic models</p>	<p>9 speed calculation</p> <p>9 calculate speed</p> <p>10 speed calculation</p> <p>13 concept of speed</p> <p>22 calculate speed</p> <p>24 quantitative understanding of Newton's second law</p> <p>25 Newton's second law</p> <p>27 calculate speed</p> <p>29 find the mechanical advantage</p> <p>56 find the mass/volume ratio</p>

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			115 Newton's second law—quantitative	
			120 Newton's second law—qualitative	
			138 the power equation	
			158 phases of the moon	
			159 properties of the moon	
			159 the moon's effect on tides on Earth	
			167 Earth and moon relationship	
			167 relationship between Earth and sun and summer and winter	
			177 concept of speed	
			199 Ohm's law	
			200 equation for Ohm's law	
			200 Ohm's law	
			227 calculating wave speeds	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.2.7.A 7	Inquiry and Design	Explain and apply scientific and technological knowledge.	<p>24 repeatability of investigations is necessary for verification</p> <p>28 asking scientific questions</p> <p>29 recognize that repeatability of investigations is necessary</p> <p>32 trace development of theories</p> <p>32 describe steps of the scientific method</p> <p>33 recognize repeatability of investigation is necessary for verification of evidence</p> <p>43 recognize that repeatability of investigation is necessary for validation of scientific evidence</p> <p>345 history of periodic table development</p> <p>360 historical developments of periodic table</p> <p>379 history of an invention</p> <p>401 scientific method</p>	<p>2 experiment to answer questions</p> <p>5 repeatability of results</p> <p>5 collaboration</p> <p>8 which ramp is the fastest?</p> <p>13 what kind of motion happens?</p> <p>18 how does friction affect motion?</p> <p>21 what happens when force is applied?</p> <p>23 how does acceleration depend on force and mass?</p> <p>26 where does the marble move the fastest?</p> <p>31 why does it look like the moon's shape changes?</p> <p>39 what are the properties of magnetism?</p> <p>43 what variables affect the pendulum?</p> <p>43 design scientific experiments</p> <p>46 how is color created?</p> <p>50 how are temperature and heat related?</p>

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
				75 what is pH?

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.2.7.B 7	Inquiry and Design	Apply process knowledge to make and interpret observations.	<p>11 featured in Ancillary Skillsheet "Significant Differences in Measurement"</p> <p>12 use and understand mass measurements</p> <p>12 length measurement</p> <p>12 measurement</p> <p>13 time measurements</p> <p>24 mass measurements</p> <p>28 interpreting observations and proposing explanations</p> <p>29 construct explanations supported by direct and indirect evidence</p> <p>30 understand sensitivity of measuring tools</p> <p>33 formulate a testable hypothesis</p> <p>34 recognizing variables</p> <p>35 length units</p> <p>35 time units</p> <p>37 measurements in English and metric units</p>	<p>2 understand measurements of time</p> <p>3 identifying cause and effect relationships</p> <p>3 identify cause and effect relationships</p> <p>4 measurements</p> <p>4 mass measurements</p> <p>5 measurement and selecting appropriate tools</p> <p>5 length measurement</p> <p>6 construct reasonable explanations supported by direct and indirect data</p> <p>6 formulate testable hypothesis</p> <p>7 develop procedure someone else could follow</p> <p>7 interpreting observations and proposing explanations</p> <p>7 test hypothesis against observations</p> <p>9 make hypothesis</p> <p>9 formulate testable hypothesis</p>

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			41 recognizing and controlling variables in observations and experiments	9 test hypothesis against observations
			44 identifying cause and effect relationships	10 recognize and control variables
			45 identifying cause and effect relationships	12 design a procedure others can follow
			54 variables	13 conduct car/ramp experiment
			61 identify cause and effect relationships—real and hypothesized	14 explain using data
			62 identifying cause and effect relationships	16 understand the sensitivity of a measuring tool
			62 use measurements of time	16 use units of force
			63 make predictions	19 formulate a testable hypothesis
			63 identifying cause and effect relationships	24 control variables
			76 measurements	24 cause and effect relationships
			81 understand and use units of force	26 make testable hypothesis
			86 understand units of force	28 units of force
			97 understand and use units of force	30 units for work
			102 design an experiment that others could follow	34 length measurements in km
			128 joule is a unit of energy	38 construct explanations based on evidence
				40 constructing explanations

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			132 units of work and energy	43 design scientific experiments
			138 units of power	43 recognizing and controlling variables
			184 measuring	49 explanations based on evidence
			192 measuring electric current	49 measuring
			193 measuring voltage	55 make predictions based on observations
			198 electrical power is measured in watts	56 measure volume with a variety of methods
			242 explanations based on observations	58 interpret observations
			270 interpreting observations and making explanations	58 conduct scientific vocabulary
			284 joules and heat energy	58 make predictions on observed data
			299 measure volume of regular and irregular objects using several methods	73 measure densities
			300 measure densities	76 interpret observations
			302 measure and compare densities	77 measurements
			302 volume of regular solids	79 mass measurements
			340 make predictions based on data	80 recognizing and controlling variables
			404 construct explanations based on data	80 interpret observations
			444 explanations based on evidence	82 make predictions based on inferences from data

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
				92 design experiment that someone else can follow
				95 measuring
				95 understanding length measurements
				96 measuring
				97 measuring
				97 length measurements
				98 measuring
				101 measurements
				101 measure volume of regular objects
				102 volume of regular objects
				102 measurements
				103 measurements
				103 measure volume of regular objects
				104 measure volume of regular objects
				105 measure volume of irregular objects
				107 mass in kg and gm
				108 mass in kg and gm

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.2.7.C 7	Inquiry and Design	Identify and use the elements of scientific inquiry to solve problems.	28 29 34 41 44 45 54 61 62 63 76 102 119 150	3 3 7 7 9 10 13 24 24 32 32 38 41 43 43
			interpreting observations and proposing explanations communication is important to science recognizing variables recognizing and controlling variables in observations and experiments identifying cause and effect relationships identifying cause and effect relationships variables identify cause and effect relationships—real and hypothesized identifying cause and effect relationships identifying cause and effect relationships line graphs sketches diagrams sketches	identifying cause and effect relationships identify cause and effect relationships interpreting observations and proposing explanations test hypothesis against observations test hypothesis against observations recognize and control variables conduct car/ramp experiment control variables cause and effect relationships communicating as essential to science sketches diagrams making sketches recognizing and controlling variables design scientific experiments

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			269 communication as essential to science	45 sketches
			270 interpreting observations and making explanations	58 conduct scientific vocabulary
			318 oral presentation of results	58 interpret observations
			318 write up results	76 interpret observations
			340 line graphs	80 recognizing and controlling variables
			364 common chemical reactions—rust	80 interpret observations
			380 writing up results	92 formal lab report
			444 reading data tables	93 communicating results is essential to science
				93 writing up results
				93 lab reports
				94 writing up results
				94 lab report
				110 lab report
				111 making graphs
				111 reading graphs

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.2.7.D 7	Inquiry and Design	Know and use the technological design process to solve problems.	7 testing models 7 design cycle 39 steps of design cycle 48 design and test a model to solve a problem 97 create and interpret dimensional drawings 99 dimensioned drawings 124 design and test a model that solves a problem 150 design a model to solve a problem 214 propose a solution to a problem 325 design and test model	45 sketch the wave fronts

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.4.10.A 10	Physical Science, Chemistry and Physics	Explain concepts about the structure and properties of matter.	<p>5 properties of solids and liquids and gas</p> <p>18 matter is composed of atoms</p> <p>19 describe matter as composed of atoms</p> <p>20 matter as composed of atoms</p> <p>20 recognize that compounds are composed of elements</p> <p>174 structure of atoms—electrons and protons and neutrons</p> <p>280 physical differences between phases of matter</p> <p>280 phases of matter</p> <p>281 phase changes</p> <p>301 physical differences between states of matter</p> <p>310 explain matter states based on arrangement of atoms</p> <p>314 compounds composed of elements</p> <p>324 atoms are made up of proton and neutron and electron</p>	<p>16 forces as ability to change motion</p> <p>52 common phase changes</p> <p>53 states of matter based on arrangement and motion of atoms</p> <p>53 phase changes</p> <p>59 understand the structure of an atom based on protons and neutrons and electrons</p> <p>59 what isotopes are</p> <p>59 atomic symbol and atomic number and mass number</p> <p>59 Bohr model</p> <p>61 periodic table</p> <p>62 understand the structure of an atom based on protons and neutrons and electrons</p> <p>62 structure of an atom</p> <p>63 identify symbols and atomic number and mass number</p> <p>63 what isotopes are</p>

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			325 protons neutrons and electrons	64 identify symbol and atomic number and mass number of elements
			326 basic properties of an atom and the three subatomic particles	64 build and describe periodic table
			328 understand how atomic structure determines the identity of elements—atomic number	65 identify groups or families on periodic table
			329 structure of an atom and three smaller particles	66 identify symbol and atomic number and mass number of elements
			330 explain what isotopes are	66 periodic table
			331 three subatomic particles and their charge	67 recognizing groups or families
			331 explain what isotopes are	67 identify metals and nonmetals and metalloids
			336 electron shells	67 periodic table
			339 structure of atoms	68 periodic table
			339 properties of subatomic materials	69 structure of an atom
			343 explain common chemical properties of elements in relation to the periodic table	69 how electrons interact to create chemical bonds
			345 common chemical properties in relation to the periodic table	70 how electrons interact to create bonds
			345 describe periodic table	70 how positive and negative ions are created
				70 structure of an atom

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			345 recognizing groups or families on the periodic table	70 three subatomic particles—charge and mass
			345 idea of atomic mass	78 how energy is manifested in chemical reactions
			346 identify metals and nonmetals on the periodic table	79 investigate law of conservation of mass
			347 explain what isotopes are	80 investigate and recognize that the chemical reactions can be represented as systems with reactants and products
			348 common chemical properties of elements based on relation to periodic table	
			348 recognizing groups and families of periodic table	80 law of conservation of mass
			348 recognizing metals and nonmetals and metalloids	81 explain how special bonding properties of carbon make possible the great variety and complexity of biomolecules
			348 describing periodic table	
			349 explain common chemical properties in relation to placement on periodic table	81 carbon and hydrogen and nitrogen and oxygen combine to form biomolecules
			350 recognizing groups and families and periodic table	
			350 describe periodic table	82 understand that carbon and hydrogen and nitrogen and oxygen combine to form biomolecules
			351 explain common chemical properties of elements in relation to periodic table	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
352			properties in relation to periodic table	
352			how electron interactions create bonds	
352			describe periodic table	
353			chemical properties in relation to periodic table	
353			groups on periodic table	
357			describe characteristics based on place in periodic table	
360			identify symbols of atoms	
363			recognize that compounds are made of elements	
364			understand biomolecules	
364			compounds are composed of elements	
365			difference between covalent and ionic bonds	
365			how electrons are involved in bonds	
365			explain chemical reactions in terms of atoms and molecules	
366			rules for writing formulas	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
368			difference between ionic and covalent compounds	
368			difference between ionic and covalent bonds	
368			how electrons are involved in bonds	
368			how ions are formed	
369			energy changes that accompany chemical reactions	
369			properties of elements in relation to the periodic table	
369			how electron interactions help create chemical bonds	
370			energy changes that accompany chemical changes	
370			distinguish between ionic and covalent compounds	
371			explain the chemical properties of elements in relation to periodic table	
371			how electrons are involved in bonds	
372			how electrons are involved in bonding	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
372			explain chemical properties based on location in periodic table	
373			how electrons are involved in bonding	
374			chemical properties based on placement in periodic table	
374			explain how ions are formed	
374			how electrons are involved in bonding	
375			apply rules for writing formulas of simple chemical compounds	
375			apply rules for writing formulas of simple chemical compounds	
376			explain why ions are formed	
376			distinguish between ionic compounds and covalent molecules	
377			qualitative understanding of how electron interactions create bonds	
392			compare covalent and ionic bonds	
396			how ions are formed	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
407			metabolism is a collection of reactions	
408			chemical reactions occur all around us	
409			chemical reactions can be represented as systems of reactants and products	
409			explain chemical changes in terms of arrangement of molecules and atoms	
410			chemical reactions in terms of atoms and molecules	
410			chemical reactions can be represented as systems with reactants and products	
410			chemical reactions in terms of arrangements of molecules and atoms	
411			write balanced chemical equations	
411			conservation of mass	
412			write balanced chemical equations	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
413			recognize that chemical reactions can be represented as systems with reactants and products	
413			balance equations	
414			explain how energy is manifested in chemical reactions—exothermic and endothermic	
415			how energy is manifested in chemical reactions	
415			analyze energy changes that accompany chemical reactions	
416			reaction types	
417			balance chemical equations	
417			common types of chemical reactions	
418			common types of chemical reactions	
419			chemical reactions occur all around us—in cars	
420			conservation of mass	
421			balanced chemical reactions	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
421			common types of chemical reactions	
421			energy manifested in chemical reactions	
422			chemical reactions all around us	
422			energy changes in chemical reactions	
423			chemical reactions all around us	
424			chemical reactions take place all around us	
427			special properties of carbon that make possible a great variety and complexity of biomolecules	
428			understand biomolecules	
428			special bonding properties of carbon	
430			metabolism as interrelated chemical reactions	
430			special bonding properties of carbon	
432			metabolism as an interrelated collection of chemical reactions	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
434			special bonding properties of carbon	
435			metabolism	
435			carbon and hydrogen and oxygen make up biomolecules	
437			different types of reactions—factors that affect reaction rates	
437			metabolism—interrelated chemical reactions	
438			metabolism	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.4.10.B 10	Physical Science, Chemistry and Physics	Analyze energy sources and transfers of heat.	10 conservation of energy (i.e. potential and kinetic)	36 series circuit
			118 Newton's third law—action and reaction	37 series circuit
			119 Newton's third law—qualitative	37 parallel circuit
			119 Newton's third law—qualitative	
			121 Newton's third law	
			122 third law and rockets	
			124 Newton's third law	
			124 third law qualitative understanding	
			130 law of conservation of energy	
			131 conservation of energy	
			134 conservation of energy in a broader context	
			192 electric current	
			193 understanding voltage	
			194 current in simple circuits	
			194 how batteries work	
			199 Ohm's law	
			199 electrical resistance	
			199 concept of voltage	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			199 concept of current	
			200 Ohm's law	
			201 resistance of common objects	
			212 current	
			212 voltage	
			213 current	
			369 energy changes that accompany chemical reactions	
			370 energy changes that accompany chemical changes	
			415 analyze energy changes that accompany chemical reactions	
			422 energy changes in chemical reactions	

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.4.10.C 10	Physical Science, Chemistry and Physics	Distinguish among the principles of force and motion. Distinguish among the principles of force and motion.	4 speed 4 Newton's second law qualitative 6 speed 17 force as ability to change motion 22 efficiency of electric motors 22 efficiency of gasoline engine 34 speed 35 understand and calculate speed 40 understand concept of speed 46 how light helps us see 46 how infrared light is used in real life 47 how infrared is used in technology 54 concept of speed 55 speed 55 compare and contrast speed and velocity 56 compare and contrast speed and velocity	8 measurement and calculation of speed 8 determine the speed of an object 9 speed calculation 9 determine speed of an object 10 speed calculation 13 concept of speed 15 calculate speed of a moving object 18 investigate the effects of friction on motion of objects 18 investigate the effects of friction on motion of objects 19 investigate effects of friction on motion 19 investigate effects of friction on motion 20 effects of friction 20 effects of friction 21 force as an action that changes motion

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			56 speed	21 balanced or unbalanced forces causing changes in motion
			60 calculating speed	
			60 changes in motion can be represented graphically	22 acceleration as change in speed
			64 calculating speed	22 calculate speed
			64 changes in motion can be shown graphically	23 Newton's second law
			64 position vs. time graphs	24 quantitative understanding of Newton's second law
			65 changes in motion can be represented graphically	24 understanding of acceleration as a rate of change of velocity
			65 speed vs. time graph	
			66 speed vs. time graph	25 Newton's second law
			66 constant velocity	27 calculate speed of a moving object
			66 calculations for speed	
			66 compare and contrast speed and velocity	28 simple machines
			67 conceptual understanding of acceleration as describing change in speed	29 efficiency of simple machines
			68 quantitative understanding of acceleration as a rate of change of velocity	30 simple machines
			69 speed vs. time graph	31 calculate speed of the car
			70 effect of gravity on motion	35 concept of electrical charge
				35 concept of electric current
				39 magnetic field

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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			70 effect of gravity on motion	40 investigate relationship between magnetism and electricity using electromagnets
			72 projectile explained	
			72 projectile explained	
			73 quantitative understanding of acceleration as change in speed	41 use pendulum to explore amplitude and period and frequency and cycle
			73 Newton's second law	42 use pendulum to investigate cycle and period
			74 how light helps us see	43 use pendulum to investigate period
			75 how light helps us see	44 reflection
			75 speed	44 demonstrate waves using slinky
			76 speed	
			76 position vs time graphs	44 wavelength and frequency and speed of waves
			79 changes in motion require force	45 making circular waves in a ripple tank
			80 understand force as an action with potential to change motion	45 categorize waves by how they move
			80 forces needed to change motion	45 wave characteristics through water
			82 effect of gravity on objects	46 RGB model of color
			82 effect of gravity on objects	46 reflected vs emitted light
			87 effect of friction on motion	46 how light helps us see
			87 effect of friction on motion	

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			88 effects of friction on motion of objects	47 investigate RGB and CMYK models of color
			88 effects of friction on motion of objects	47 RGB and CMYK color models
			89 effects of friction on motion	48 refractive optics such as lenses
			89 effects of friction on motion	48 refraction of light
			89 changes in motion require application of force	
			90 effects of friction on motion	
			90 effects of friction on motion	
			91 effects of friction	
			91 effects of friction	
			92 effects of friction	
			92 effects of friction	
			93 effects of friction on motion	
			93 effects of friction on motion	
			94 compare and contrast constant and changing velocity	
			95 balanced and unbalanced forces	

**Correlation to Pennsylvania Academic Standards for Science
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Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
96			effects of gravity	
96			effects of gravity	
96			change in motion require force	
97			use concepts of balanced or unbalanced forces	
97			quantitative understanding of force changing motion	
98			balanced and unbalanced forces	
99			unbalanced forces cause motion	
99			effect of friction on motion	
99			effect of friction on motion	
100			Newton's laws applied to real situations	
100			how gravity affects motion	
100			Newton's first law	
100			how gravity affects motion	
101			Newton's laws applied to planes	
101			concept of forces	
101			effects of friction (drag) on motion of planes	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
101			effects of friction (drag) on motion of planes	
102			investigate how friction affects motion	
102			investigate how friction affects motion	
108			conceptual understanding of a force as the action with the potential to change motion	
108			changes in motion require force	
109			change in motion requires force	
110			compare and contrast constant and changing velocity	
111			force is an action with potential to change motion	
113			acceleration is a rate of change of speed	
113			concept of acceleration	
113			Newton's second law	
114			Newton's second law—qualitative	
114			Newton's second law—qualitative	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
114			Newton's second law—qualitative	
115			acceleration is a change in speed	
115			Newton's second law—quantitative	
116			conceptual understanding of acceleration as change in speed	
117			understand and use concept of balanced and unbalanced forces to create motion	
118			Newton's third law—action and reaction	
118			balanced and unbalanced forces	
119			Newton's third law—qualitative	
119			Newton's third law—qualitative	
120			Newton's laws in terms of real situations—sports and cars	
120			Newton's second law—qualitative	
121			effect of friction on objects	
121			effect of friction on objects	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
121			Newton's third law	
122			third law and rockets	
123			Newton's laws and space travel	
124			force as ability to change motion	
124			Newton's second law	
124			Newton's first law	
124			Newton's third law	
124			third law qualitative understanding	
136			understanding of force as the ability to change motion	
140			simple machines	
141			simple machines and forces	
142			friction and machines	
142			friction and machines	
143			efficiency explained	
144			simple machines and the human body	
144			how a lever works	
145			how a rope and pulley system works	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
146			how gears and ramps work	
156			effect of gravity on motion	
156			effect of gravity on motion	
157			constant and changing velocity	
160			effect of gravity on motion of objects	
160			effect of gravity on motion of objects	
177			concept of speed	
190			compare and contrast electrical force and magnetic force	
190			electric charge	
191			understanding electric charge	
191			what causes shocks	
191			charged objects and static electricity	
203			what is a magnet	
204			concept of magnetic field	
206			magnetic vs electric force	
207			finding the poles of an electromagnet using right-hand rule	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
209			electromagnetic induction explained	
214			how coils concentrate magnetic field	
214			relationship between magnetic field and electric charge	
218			understanding a cycle	
218			a pendulum's cycle	
220			period is the time for one cycle	
220			frequency is the inverse of period	
220			frequency explained	
221			friction and damping	
221			friction and damping	
221			amplitude explained	
223			resonance explained	
225			waves transmit energy	
226			frequency and amplitude and wavelength of waves	
227			the speed of waves	
228			longitudinal waves	
228			transverse waves	
229			reflected waves	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			229 refracted waves	
			230 destructive interference	
			233 acoustics	
			234 speed of sound	
			235 sound as a wave	
			235 electromagnetic waves in common technology (i.e. radar)	
			235 wavelength and frequency	
			236 sound as a wave	
			237 how a French horn works	
			237 wavelength of sound	
			238 how the ear works	
			240 reflection of waves	
			240 why sound is a wave	
			240 how we hear sound	
			240 absorption of sound waves	
			241 how we hear sound	
			242 sound as a wave	
			242 how we hear sound	
			246 seeing and reflected light	
			247 light is produced by atoms	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
248			white light is a mixture of colors	
249			speed of light	
249			speed of light	
250			energy and color of light	
250			wavelength and frequency of visible light	
251			nature of light in terms of waves and energy info flow	
251			nature of light in terms of waves and energy info flow	
251			what makes light	
252			uses of electromagnetic waves	
252			properties of electromagnetic waves with different wavelengths	
252			electromagnetic spectrum	
254			how the human eye sees light	
255			how the human eye sees color	
256			the RGB color process	
257			color as reflected light	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
257			subtractive color process	
257			how light helps us see	
258			the CMYK color process	
259			explain how colors of light relate to wavelength	
259			color seen as reflected light	
261			mirrors reflect light	
262			how light helps us see	
262			wave interactions like reflection	
262			absorption	
262			refraction	
263			reflection of light	
263			refraction of light	
264			reflection explained	
268			what makes light	
299			investigate density of fluid	
305			investigate buoyancy	
306			investigate properties—including buoyancy	
309			investigate buoyancy	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
332			explain how we see color in terms of reflected or emitted light	
390			factors that affect solubility	

Correlation to Pennsylvania Academic Standards for Science CPO Science *Physical Science (Middle School)*

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.4.10.D 10	Physical Science, Chemistry and Physics	Explain essential ideas about the composition and structure of the universe.	156 orbits of planets and moons and other bodies 157 describe orbits of planets 158 orbit of moon 160 place of Earth in the solar system 161 general structure of the solar system 162 general position of Earth 163 general structure of solar system 164 orbits of other bodies in the solar system 164 other bodies in solar system 165 general structure of solar system 166 compare orbits of planets 169 orbits of moons 170 orbits of other moons 171 compare orbits of planets in solar system 171 theories of origin of universe 173 orbits on bodies in solar system	32 orbit of the moon 33 general characteristics of the solar system 34 know general structure of the solar system 34 relationship of the solar system to the universe—Milky Way galaxy

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
176			explain how stars form and features of life cycle	
176			compare the Sun to other stars	
176			explain how stars form and features of life cycle	
177			explain use of light years	
178			how stars produce energy	
178			explain how stars produce energy	
178			how stars produce energy	
178			explain how stars produce energy	
179			relationship of solar system to the universe	
179			general characteristics of universe—galaxies	
179			light years	
180			light years	
180			characteristics of the universe	
181			description of galaxy as we know it	
181			light years	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.4.7.A 7	Physical Science, Chemistry, and Physics	Describe concepts about the structure and properties of matter.	18 matter is composed of atoms 19 describe matter as composed of atoms 20 matter as composed of atoms 20 recognize that compounds are composed of elements 174 structure of atoms—electrons and protons and neutrons 314 compounds composed of elements 316 classify materials based on physical properties 317 classify materials based on properties 324 atoms are made up of proton and neutron and electron 329 structure of an atom and three smaller particles 339 structure of atoms 344 chemical and physical properties	59 understand the structure of an atom based on protons and neutrons and electrons 62 understand the structure of an atom based on protons and neutrons and electrons 62 structure of an atom 69 structure of an atom 70 structure of an atom 77 difference between chemical and physical changes 80 investigate and recognize that the chemical reactions can be represented as systems with reactants and products

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
344			difference between chemical and physical changes	
350			physical and chemical properties	
357			classify by chemical and physical properties	
357			melting and boiling points	
358			physical properties of gold	
363			recognize that compounds are made of elements	
364			compounds are composed of elements	
368			difference between ionic and covalent bonds	
407			differences between chemical and physical change	
408			chemical reactions occur all around us	
408			difference between physical and chemical change	
409			chemical reactions can be represented as systems of reactants and products	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
410			chemical reactions can be represented as systems with reactants and products	
411			write balanced chemical equations	
412			write balanced chemical equations	
413			recognize that chemical reactions can be represented as systems with reactants and products	
413			balance equations	
413			difference between chemical and physical change	
416			reaction types	
417			balance chemical equations	
417			common types of chemical reactions	
418			common types of chemical reactions	
419			chemical reactions occur all around us—in cars	
421			common types of chemical reactions	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
421			balanced chemical reactions	
422			chemical reactions all around us	
423			chemical reactions all around us	
424			chemical reactions take place all around us	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.4.7.B 7	Physical Science, Chemistry, and Physics	Relate energy sources and transfers to heat and temperature.	5 conversion of energy types 6 energy conversions 10 conversion of energy	26 transformation of energy from one form to another 27 transformation of energy from one form to another
			15 conversion of energy from one form to another	35 construct simple circuits
			15 types and forms of energy	36 series circuit
			16 conversion of energy from one form to another	36 simple circuits
			31 basic forms of energy—heat	37 parallel circuit
			46 how infrared light is used in real life	37 series circuit
			47 how infrared is used in technology	37 simple circuits
			130 basic forms of energy	
			131 conversion of energy from potential to kinetic	
			148 forms of energy (i.e. wind)	
			149 conversions of energy	
			150 energy conversions	
			196 electric circuits	
			235 Doppler effect	
			248 white light is a mixture of colors	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			250 wavelength and frequency of visible light 252 electromagnetic spectrum	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.4.7.C 7	Physical Science, Chemistry, and Physics	Identify and explain the principles of force and motion.	4 speed	8 determine the speed of an object
			6 speed	8 measurement and calculation of speed
			34 speed	
			35 understand and calculate speed	9 determine speed of an object
			40 understand concept of speed	9 speed calculation
			54 concept of speed	10 speed calculation
			55 compare and contrast speed and velocity	13 concept of speed
			55 speed	15 calculate speed of a moving object
			56 speed	22 calculate speed
			56 compare and contrast speed and velocity	27 calculate speed of a moving object
			60 calculating speed	31 calculate speed of the car
			64 position vs. time graphs	44 wavelength and frequency and speed of waves
			64 calculating speed	44 wave as oscillation in a medium
			65 speed vs. time graph	
			66 speed vs. time graph	
			66 calculations for speed	
			66 compare and contrast speed and velocity	
			69 speed vs. time graph	
			75 speed	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
			76 speed	
			76 position vs time graphs	
			177 concept of speed	
			225 waves transmit energy	
			226 frequency and amplitude and wavelength of waves	
			227 the speed of waves	
			234 speed of sound	
			235 sound as a wave	
			235 wavelength and frequency	
			236 sound waves and different media	
			236 sound as a wave	
			237 wavelength of sound	
			240 why sound is a wave	
			242 sound as a wave	
			254 how the human eye sees light	
			261 mirrors reflect light	
			263 reflection of light	
			264 reflection explained	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
3.4.7.D 7	Physical Science, Chemistry, and Physics	Describe essential ideas about the composition and structure of the universe and the earth's place in it.	156 orbits of planets and moons and other bodies 157 describe orbits of planets 157 role of gravity in solar system 158 orbit of moon 158 role of gravity in universe 158 moon reflects light 164 other bodies in solar system 164 orbits of other bodies in the solar system 164 how astronomical instruments help us understand the universe 165 compare Earth to other planets with respect to supporting life 166 classify and compare planets 166 astronomical instruments help us understand units 166 compare orbits of planets 166 compare planets to Earth 167 classify and compare planets	32 orbit of the moon

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
168			classify planets	
168			compare other planets with Earth with respect to supporting life	
169			classify planets	
169			orbits of moons	
169			compare other planets to Earth with respect to supporting life	
170			classify and compare planets	
170			orbits of other moons	
171			theories of origin of universe	
171			compare orbits of planets in solar system	
171			compare and classify planets	
171			how astronomical instruments help us understand universe	
172			classify planets	
172			astronomical instruments help us understand universe	
173			orbits on bodies in solar system	

**Correlation to Pennsylvania Academic Standards for Science
CPO Science Physical Science (Middle School)**

Standard #: by end of grade	Category	Standard Statement	Volume 1 Student Text page	Volume 2 Investigation Manual page
173			astronomical instruments used to understand universe	
173			classify planets	
173			compare Earth to other planets with respect to supporting life	
175			explain and diagram the structure of the Sun	
176			compare the Sun to other stars	
176			explain how stars form and features of life cycle	
178			explain how stars produce energy	
178			how stars produce energy	
182			how astronomical instruments increase our understanding of universe	
183			how astronomical instruments have increased our knowledge of universe	
184			structure of sun	
286			Earth compared with other planets with respect to supporting life	