

Correlation to District of Columbia Science Standards: Essential Skills
***Foundations of Physical Science* Student Text and Investigation Manual**

Standard #: Grade	Subject	Content Standard	Essential Skills	student text pg	detail	investigation pg	detail
PS.3.01 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student draws models that show the structure and parts of an atom	311	location/size/charge of subatomic particles	132	atomic number determines what element that atom is
				311	protons/neutrons/electrons	132	building atom models
				315	atoms of same element have same atomic number	133	location of electrons in atom
				318	proton/electron attraction	133	protons and neutrons
				388	showing valence electrons in a diagram	136	model stable and neutral atoms
						137	importance of atomic number
						137	build atomic models
						140	find the number of electrons in outermost level
						140	review subatomic particles

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PS.3.02 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student observes, describes, and measures physical properties of matter including melting point, boiling point, solubility, mass, volume and density	278	mixtures can be separated by physical means	114	separating a homogeneous mixture
				281	volume and mass contrasted	116	mass and volume measurements
				284	melting and boiling point explained	118	observe melting process and study quantitatively
				284	changes of state	119	melting point of ice
				284	melting and boiling points	119	create a temperature vs. time graph of phase change
				285	table of melting and boiling points	119	energy and phase changes
				285	characteristics of matter related to its state	124	build a density column
				291	density is independent of amount of substance	126	investigating buoyancy with clay boats
				291	density explained	128	use CPO viscometer to study viscosity
				292	elasticity is a physical property of matter	166	design experiments to explore dissolving rate
				292	hardness is a physical property of matter	167	what happened at the molecular level?
				293	brittleness is a physical property of matter	167	investigate the dissolving process
				294	development of Kevlar brand fiber	168	investigate solubility of sugar
				294	malleability is a physical property of matter	170	solubility and temperature
				294	tensile strength is a physical property of matter		

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				295	relationship between mass volume and density		
				296	density of liquid water vs. ice		
				297	buoyancy explained		
				298	sinking and floating		
				302	viscosity of motor oils		
				305	viscosity of glue mixtures		
				406	molecular motion and dissolving rate		
				406	molecular motion and dissolving rate		
				407	surface area and dissolving rate		
				411	effect of temperature on solubility		
				421	why water is a nearly universal solvent		
				423	polar solutes		
PS.3.03 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student differentiates mass and weight	47	weight vs. mass	20	weight vs. mass
				53	how to calculate weight	24	measure force in newtons
				69	newtons and pounds		

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PS.3.04 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student explains how temperature affects the average kinetic energy of atoms	451 452	temperature and kinetic energy temperature scales	119	investigate temperature and energy transfer in melting process
PS.3.05 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student illustrates how changes in energy causes change in phase	284 284	states of matter and arrangement of molecules changes of state	118 118 119 119 119	molecules in a liquid investigate melting investigate melting and create a graph create a temperature vs. time graph of phase change energy and phase changes
PS.3.06 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student differentiates physical and chemical changes in matter	294 353 355 372	development of Kevlar brand fiber physical and chemical changes and digestion physical and chemical changes in tire recycling determine if changes are chemical or physical	146	investigate and observe chemical and physical changes in the lab

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PS.3.07 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student determines relationships between mass and volume	78	use and understand mass measurements	18	use a balance to find mass of car
				280	measuring volume of solids	116	measuring mass
				280	measuring volume of liquids	116	mass and volume measurements
				281	volume and mass contrasted	117	measuring volume
				295	relationship between mass volume and density		

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PS.3.08 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student compares and contrasts the properties of particles and waves	195	waves transmit energy	94	does sound behave like other waves?
				222	effect of medium on speed of sound wave	132	comparing atoms
				222	effect of temperature on speed of sound wave	140	find the number of electrons in outermost level
				311	all matter is formed from atoms	141	modeling a chemical bond
				311	all matter is formed from atoms		
				324	use the periodic table to predict chemical formulas		
				324	which element is more likely to combine with other elements?		
				335	chemical bonding and the periodic table		
				388	showing valence electrons in a diagram		
				474	energy and radiation relationships		

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PS.3.09 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student describes the general properties of atoms and how these properties change when electrons are transferred from one substance to another	315	atoms of same element have same atomic number	132	atomic number determines what element that atom is
				324	use the periodic table to predict chemical formulas	136	ions
				324	which element is more likely to combine with other elements?	137	importance of atomic number
				330	Lewis Dot diagrams	141	whan an atom ionizes
				335	chemical bonding and the periodic table	141	modeling a chemical bond
				354	new substances are formed when a chemical change occurs	157	predict the products of double displacement reactions
				357	chemical reactions involve rearrangement of atoms	161	research pros and cons of uses for radioactive elements
				364	carbon chains	162	carbon reactions and the environment
				393	carbon dating		
				393	radioisotopes in science and medicine		
				400	research pros and cons of nuclear technology		

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PS.3.10 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student describes forms of radiant, light and thermal energy and their applications to everyday life	91	following an energy transformation	39	make an energy flow chart
				91	understand basic forms of energy	39	identify type of energy involved
				195	waves transmit energy	119	investigate temperature and energy transfer in melting process
				468	heat transfer through air	147	feel the heat generated by chemical reaction
				468	densely packed solids are good conductors of heat	190	investigate conduction through all states of matter
				470	convection currents and weather	192	investigate convection in liquids
				470	warming hands over candle	194	investigate radiation emitted by solids
				472	convection currents in water	194	investigate radiation emitted by liquids
				474	energy and radiation relationships		
				476	solid road surface emits radiation		
				478	apply knowledge of heat transfer to different situations		
				481	global warming and heat transfer by radiation		

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PS.3.11 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student gives examples of how different types of energy can be transferred by radiation, conduction and convection	468 468 470 470 472 476 478 481	heat transfer through air densely packed solids are good conductors of heat convection currents and weather warming hands over candle convection currents in water solid road surface emits radiation apply knowledge of heat transfer to different situations global warming and heat transfer by radiation	190 192 194 194	investigate conduction through all states of matter investigate convection in liquids investigate radiation emitted by solids investigate radiation emitted by liquids
PS.3.12 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student explains how the principle of conservation of energy is applied during an energy transfer	88 90 91 92 93 96	potential and kinetic energy explained conservation of energy explained energy conversions energy transformations and conservation different forms of energy described prove that energy is conserved	36 37 38 38 188	energy conservation and the roller coaster investigating conservation of energy with rollercoaster explore energy transformations conservation of energy and energy transformations specific heat and conservation of energy

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PS.3.13 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student explains properties and parts of waves, and the effects of different media on waves	179	what is a cycle?	83	measure speed of a wave pulse
				182	concept of period explained	83	find speed of a wave
				182	concept of frequency explained	85	observing reflection in water waves
				192	analyze systems to find cycle/period/frequency	86	investigate frequency and wavelength
				195	waves transmit energy	86	adjust frequency of a standing wave
				198	frequency and wavelength and amplitude	95	interference and sound waves
				201	waves and refraction	108	explore refraction with a prism
				201	reflection in water waves and light waves		
				201	waves and reflection		
				201	waves and absorption		
				202	refraction and eyeglasses		
				215	properties of sound waves		
				221	importance of wavelength of sound waves		
				242	properties of light waves		
				261	refraction and lenses		
				474	energy and radiation relationships		
				474	electromagnetic radiation		

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PS.3.14 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student defines, estimates and calculates speed, velocity and acceleration of objects	14	how to calculate speed	8	calculating speed
				15	compare and contrast speed and velocity	9	collect data and calculate speed of car
				20	find speed of bumblebee	10	calculate speed of the car
				20	calculate speed of car	12	calculate speed of moving car
				24	accurate speed measurements	12	find speed of car at different positions
				30	position vs. time graphs	13	make a position vs. time graph
				32	average speed vs. instantaneous	14	calculate acceleration of car on ramp
				32	average speed discussed	14	acceleration is the rate at which speed changes
				33	understanding acceleration	14	calculate speed of car at two places on the ramp
				35	how to calculate acceleration	15	make a speed vs. time graph
				36	examples of acceleration	17	explore 2nd law and acceleration
				37	speed vs. time graphs	17	calculate speed of car
				41	find acceleration of car	36	find speed of marble
				42	calculate speed from distance/time graph		
				49	link between force and acceleration		
				53	acceleration due to gravity		

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PS.3.15 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student measures the frequency of waves and identifies devices that use wave energy	179	what is a cycle?	83	measure speed of a wave pulse
				182	concept of frequency explained	83	find speed of a wave
				182	concept of period explained	86	investigate frequency and wavelength
				192	analyze systems to find cycle/period/frequency	86	adjust frequency of a standing wave
				195	waves transmit energy		
				198	frequency and wavelength and amplitude		
				215	properties of sound waves		
				221	importance of wavelength of sound waves		
				242	properties of light waves		
				474	energy and radiation relationships		

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PS.3.16 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student describes the frequency of waves and identifies devices that use wave energy	179 182 182 192 195 198 215 221 242 474	what is a cycle? concept of frequency explained concept of period explained analyze systems to find cycle/period/frequency waves transmit energy frequency and wavelength and amplitude properties of sound waves importance of wavelength of sound waves properties of light waves energy and radiation relationships	83 83 86 86	measure speed of a wave pulse find speed of a wave investigate frequency and wavelength adjust frequency of a standing wave
PS.3.17 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student describes the properties of forces that act at a distance (e.g., gravitational, electrical, magnetic)	52 54 55 106 106 159 163	gravity depends on mass Newton's law of universal gravitation calculating gravitational force between objects electrical force is incredibly strong! electrical forces magnetism explained understanding magnetic fields	62 64 66	describing forces that magnets exert on each other testing materials to see if they are affected by magnets compare electromagnets and permanent magnets

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PS.3.18 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student constructs an electromagnet then explains the forces at work	164 166 166 171	what is an electromagnet? increased current vs. strength of magnetic field building an electromagnet electromagnetic induction explained	66 67 73 73	build an electromagnet find out what happens to strength of electromagnet when current is increased exploring electric generators use magnetic induction to create an electric field
PS.3.19 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student differentiates parallel and series circuits	145 145 145 145 146 147 151 155 156	single path vs. branching paths holiday lights as series or parallel parallel circuit defined series circuit defined household wiring current and voltage in series circuits voltage and resistance in parallel circuits analyze a parallel circuit analyze a series circuit	56 56 57 58 60 61	build a parallel circuit build a series circuit compare brightness of bulbs in series vs. parallel build a series circuit and find total resistance parallel circuit and Ohm's law compare current and voltage and resistance in each type of circuit

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PS.3.20 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student describes some common contact forces(e.g., buoyancy, tension, friction)	56 64 291 292 292 293 294 294 297 298	friction explained research effect of friction on human joints density is independent of amount of substance elasticity is a physical property of matter hardness is a physical property of matter brittleness is a physical property of matter tensile strength is a physical property of matter malleability is a physical property of matter buoyancy explained sinking and floating	21 126	effect of friction on the car investigating buoyancy with clay boats
PS.3.21 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student identifies forces that act on an object and the effect of the forces on the object	51 51	net force explained balanced and unbalanced forces	16 22	unbalanced forces and acceleration of car car and ramp and Newton's 3rd law

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PS.3.22 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student explains how simple machines are used to change the direction or size of a force	67	how simple machines manipulate forces	12	calculate speed of moving car
				69	how to calculate mechanical advantage	25	discover mechanical advantage of ropes and pulleys
				70	mechanical advantage of block and tackle	27	changing force and distance on a lever
				71	the human body and simple machines	27	set up a lever that has mechanical advantage
				71	parts of a lever	30	exploring force and distance with ropes and pulleys
				71	how a lever works		
				71	pliers as an example of a lever		
				72	mechanical advantage of a lever		
				75	how gears work		
				78	set up a lever with MA greater than 1		
				78	design a toothbrush		
				79	calculate mechanical advantage		
				79	analyze pulleys with different numbers of supporting strings		
				79	analyze block and tackle		
				80	analyzing the jaw as a lever		
				80	analyze block and tackle machine on a sailboat		

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				80	analyze wheelbarrow		
PS.3.23 8	Physical Science	Each student describes the structure of matter, the physical and chemical changes it undergoes, and its energy sources and transformations.	The student explains how gravity is a force that every mass exerts on every other mass	52 52 54 55	the effect of gravity gravity depends on mass Newton's law of universal gravitation calculating gravitational force between objects	20	investigate effect of gravity on motion